

mólová hmotnosť látky L (hmotnosť m_L na 1 mól)	M_L	kg mol ⁻¹
mólový objem látky L (objem V_L na 1 mól)	$V_{m,L}$	m ³ mol ⁻¹
Avogadrova konštanta (počet častíc N_L na 1 mól)	$N_A = 6,02214076 \cdot 10^{23} \text{ mol}^{-1}$ (presne)	mol ⁻¹
látkové množstvo látky L	$n_L = \frac{m_L}{M_L} = \frac{V_L}{V_{m,L}} = \frac{N_L}{N_A}$	mol
hustota čistej látky L	$\rho_L = \frac{m_L}{V_L}$	kg m ⁻³
hmotnosť jednej častice	$m_{1L} = \frac{m_L}{N_L} = \frac{M_L}{N_A}$	kg
atómová hmotnostná jednotka	$u = 1,66053906660 \cdot 10^{-27} \text{ kg}$	kg
hustota zmesi	$\rho' = \frac{m'}{V'}$	kg m ⁻³
hmotnostný zlomok látky L v zmesi	$w_L = \frac{m_L}{m'} \quad \sum_i w_i = 1$	1
objemový zlomok látky L v zmesi	$\phi_L = \frac{V_L}{V'} \quad \sum_i \phi_i = 1$ *	1
mólový zlomok látky L v zmesi	$x_L = \frac{n_L}{n'} \quad \sum_i x_i = 1$	1
hmotnostná koncentrácia látky L v zmesi	$\rho_L = \frac{m_L}{V'} \quad \sum_i \rho_i = \rho'$	kg m ⁻³
látková koncentrácia (molarita) látky L v zmesi	$c_L = \frac{n_L}{V'}$	mol m ⁻³ , mol dm ⁻³ \equiv M
molarita látky L v zmesi	$\underline{m}_L \equiv b_L = \frac{n_L}{m_S}$ ** $\underline{m}_S \equiv b_S = \frac{n_S}{m_S} = \frac{1}{M_S}$	mol kg ⁻¹

štandardné podmienky	$p^0 = 10^5 \text{ Pa}, T^0 = 298,15 \text{ K}$	
štandardný mólový objem ideálneho plynu L	$V_{m,L(g)}^0 = 24,78957 \text{ dm}^3 \text{ mol}^{-1}$	m ³ mol ⁻¹
univerzálna plynová konštanta	$R = 8,31446261815324 \text{ J mol}^{-1} \text{ K}^{-1}$	J mol ⁻¹ K ⁻¹
stavová rovnica ideálneho plynu L	$p_L V_L = n_L R T_L$	Pa, m ³ , mol, J, K
priemerná mólova hmotnosť zmesi	$\bar{M} = \frac{\sum_i x_i M_i}{\sum_i x_i} = \sum_i (x_i M_i)$ $\bar{M} = \frac{\sum_i w_i}{\sum_i \left(\frac{w_i}{M_i}\right)} = \frac{1}{\sum_i \left(\frac{w_i}{M_i}\right)}$ $\bar{M} = \frac{\sum_i c_i M_i}{\sum_i c_i}$ $\bar{M} = \frac{\sum_i \rho_i}{\sum_i \left(\frac{\rho_i}{M_i}\right)} = \frac{\rho'}{\sum_i \left(\frac{\rho_i}{M_i}\right)}$ $\bar{M} = \frac{\sum_i \phi_i \rho_i}{\sum_i \left(\frac{\phi_i \rho_i}{M_i}\right)} = \frac{\rho'}{\sum_i \left(\frac{\phi_i \rho_i}{M_i}\right)}$ $\bar{M} = \frac{\sum_i m_i M_i}{\sum_i m_i}$	kg mol ⁻¹
rozsah chemickej reakcie	$d\xi = \frac{dn_L}{\nu_L} \Leftrightarrow \Delta\xi = \frac{\Delta n_L}{\nu_L}$ ***	mol

* platí len pre ideálne zmiešavanie, tj. keď $\sum V_i = V'$.

** L = Látka, ktorej sa veličina týka, S = rozpúšťadlo (Solvent).

*** ν_L = stechiometrický koeficient látky L je pre reaktanty záporný, pre produkty kladný.

	w_L	φ_L	x_L	$\underline{\rho}_L$	c_L	\underline{m}_L
$w_L =$	=	$w_L = \frac{\varphi_L \rho_L}{\sum_i (\varphi_i \rho_i)}$ $w_L = \frac{\varphi_L \rho_L}{\rho'}$	$w_L = \frac{x_L M_L}{\sum_i (x_i M_i)}$ $w_L = \frac{x_L M_L}{\bar{M}}$	$w_L = \frac{\rho_L}{\rho'}$	$w_L = \frac{c_L M_L}{\rho'}$	$w_L = \frac{\underline{m}_L M_L}{\sum_i (\underline{m}_i M_i)}$ $w_L = \frac{\underline{m}_L M_L}{\bar{M} \sum_i \underline{m}_i}$ $\underline{m}_S = \frac{1}{M_S}$
$\varphi_L =$	$\varphi_L = \frac{w_L \rho'}{\rho_L}$	=	$\varphi_L = \frac{x_L M_L \rho'}{\rho_L \sum_i (x_i M_i)}$ $\varphi_L = \frac{x_L M_L \rho'}{\rho_L \bar{M}}$	$\varphi_L = \frac{\rho_L}{\rho_L}$	$\varphi_L = \frac{c_L M_L}{\rho_L}$	$\varphi_L = \frac{\underline{m}_L M_L \rho'}{\rho_L \sum_i (\underline{m}_i M_i)}$ $\varphi_L = \frac{\underline{m}_L M_L \rho'}{\rho_L \bar{M} \sum_i \underline{m}_i}$
$x_L =$	$x_L = \frac{w_L}{M_L \sum_i \left(\frac{w_i}{M_i} \right)}$ $x_L = \frac{w_L \bar{M}}{M_L}$	$x_L = \frac{\varphi_L \rho_L}{M_L \sum_i \left(\frac{\varphi_i \rho_i}{M_i} \right)}$ $x_L = \frac{\varphi_L \rho_L \bar{M}}{M_L \rho'}$	=	$x_L = \frac{\rho_L}{M_L \sum_i \left(\frac{\rho_i}{M_i} \right)}$ $x_L = \frac{\rho_L \bar{M}}{\rho' M_L}$	$x_L = \frac{c_L}{\sum_i c_i}$	$x_L = \frac{\underline{m}_L}{\sum_i \underline{m}_i}$
$\underline{\rho}_L =$	$\underline{\rho}_L = w_L \rho'$	$\underline{\rho}_L = \varphi_L \rho_L$	$\underline{\rho}_L = \frac{x_L M_L \rho'}{\sum_i (x_i M_i)}$ $\underline{\rho}_L = \frac{x_L M_L \rho'}{\bar{M}}$	=	$\underline{\rho}_L = c_L M_L$	$\underline{\rho}_L = \underline{m}_L M_L \rho_S$
$c_L =$	$c_L = \frac{w_L \rho'}{M_L}$	$c_L = \frac{\varphi_L \rho_L}{M_L}$	$c_L = \frac{x_L \rho'}{\sum_i (x_i M_i)}$ $c_L = \frac{x_L \rho'}{\bar{M}}$	$c_L = \frac{\rho_L}{M_L}$	=	$c_L = \frac{\underline{m}_L \rho'}{\sum_i (\underline{m}_i M_i)}$ $c_L = \frac{\underline{m}_L \rho'}{\bar{M} \sum_i \underline{m}_i}$
$\underline{m}_L =$	$\underline{m}_L = \frac{w_L}{M_L w_S}$	$\underline{m}_L = \frac{\varphi_L \rho_L}{M_L \varphi_S \rho_S}$	$\underline{m}_L = \frac{x_L}{M_S x_S}$	$\underline{m}_L = \frac{\rho_L}{M_L \rho_S}$	$\underline{m}_L = \frac{c_L}{\rho' - \sum_{i \neq S} (c_i M_i)}$	=

L = Látka, ktorej sa veličina týka, S = rozpúšťadlo (Solvent).